STEREOSELECTIVE SYNTHESIS OF (S)-PROPANOL AMINES: LIPASE CATALYZED OPENING OF EPOXIDES WITH 2-PROPYLAMINE

Ahmed Kamal*, Yalamati Damayanthi and Maddamsetty V. Rao

Division of Organic Chemistry

Indian Institute of Chemical Technology, Hyderabad 500 007, India

(Received 27 August 1992; accepted 23 September 1992)

Abstract: The ring opening of epoxides with 2-propylamine in presence of lipase in organic solvents affords selectively (S)-propanol amines. The effect of solvents towards selectivity has also been examined.

It is well established that in the beta-adrenergic blocking agents the biological activity resides mainly in the (S)-enantiomer¹⁻³. On the contratry the opposite (R)-enatiomer may be responsible for the side-effects⁴. As a result, in the recent years, there have been considerable efforts in the preparation of the enantiomerically pure form of (S)-propanolamines by asymmetric synthesis⁵ or bioconversions⁶. The conventional method for the synthesis of these compounds is by the opening of epoxides with corresponding amines. Based on our previous interest on the application of enzymes as biocatalysts⁷, we recently studied the stereoselective opening of epoxides by arylamines in presence of liver microsomes⁸. In continuation of this approach it was considered worthwhile to open the epoxides by an amine in presence of a lipase. This could offer an advantage of carrying out the reactions in organic media and thus enhancing the suitability of this reaction for the water-susceptible or insoluble substrates. However, lipases have been extensively employed in the enantio- or regioselective transformations⁹ involving acylations, transesterifications, hydrolyses, lactonization, hydrazinolysis, aminolysis and transamidation.

This report describes for the first time the enantioselective opening of epoxides (1) by 2-propylamine (2) in presence of porcine pancreatic lipase (PPL)¹⁰ in organic solvents. Furthermore, the effect of solvents on the selectivity of propanol amines (3) has been studied. In a typical reaction; to 1a (50 mg) and PPL (100mg) in hexane (40 ml) was added slightly more than half a molar equivalent amount of 2a. After being stirred at room temperature for 12h, the lipase is filtered and the resulting solution is

1362 A. KAMAL et al.

evaporated to give the residue. This was immediately subjected to column chromatography, chloroform-methanol (9.8:0.2) to recover the unreacted epoxide 1a. After the complete recovery of 1a the eluent was changed to chloroform-methanol (9.5:0.5) for obtaining the propanol amine (3a). The reactions were monitored by TLC.

The comparative analysis of the results in Table 1 clearly reflects the considerable influence of the solvents on the enantioselectivities of 3. With the exeception of the reaction of 1c in hexane the ring opening to (S)-form seems to be preferential in presence of lipase, particularly employing toluene as the solvent. The absolute configuration was obtained by the correlation of the reaction of 3-(1-napthyloxy)-1,2-epoxypropane with isopropyl amine to give propranolol, which was then compared with chirally pure (R)-and (S)-propranolol.

The strategy descibed here provides a facile synthesis of (S)-propanol amines by ring-opening of epoxides in presence of lipase. Thus, this methodology is noteworthy as it offers the preparation as well as resolution of propanol amines in one-pot.

Table 1. Lipase-catalyzed opening of epoxides	with 2-propylamine
---	--------------------

Entry	Ar	Reaction medium	Conversion (%)	Enantiomeric ratio (%) ^a	
				(R)	(S)
1a	C ₆ H ₅	hexane	48	39	61
		toluene	42	0	>99
		ethyl acetate	36	40	60
1b	4-ClC ₆ H ₄	hexane	41	48	52
		toluene	45	7	93
		ethyl acetate	39	19	81
1c	2-OCH ₃ C ₆ H ₄	hexane	35	62	38
		toluene	39	35	65
		ethyl acetate	43	20	80
1d	4-NHCOCH3C6H4	hexane	b	-	_
		toluene	47	0	>99
		ethyl acetate	40	20	80

aCompounds 3 were derivatized to oxazolidones 11 and enantiomeric ratio was determined by HPLC employing LKB enantiopac, α -AGP 10 μ m cartridge column (4.0 x 100 mm), 2-propanol (10% for 3a; 15% for 3c; 20% for 3b and 3d) in 8mM sodium dihydrogen phosphate / sodium hydrogen phosphate buffer with 0.1M sodium chloride pH 6.9 at 0.25 ml / min flow and 230 nm wavelength; calibrated by the authentic racemic forms of 3. bSubstrate almost insoluble in hexane.

Acknowledgements: We are grateful to Dr A V Rama Rao, Director for his interest and encouragement. One of us (YD) is thankful to UGC, New Delhi for the award of a research fellowship.

References and notes

- 1. R. Howe and R. G. Shanks, Nature, 1966, 1336.
- 2. In 'Handbook of Sterioisomers: Therapeutic Drugs', ed. Smith, D. F.; CRC Press, Florida, 1989.
- 3. A. K. Scott, Drug News & Perspectives, 1991, 4, 476.
- 4. F. Jamali, R. Mehvar and F. M. Pasutto, J. Pharm. Sci., 1989, 78, 695.

1364 A. Kamal et al.

- H. Takahashi, S. Sakuraba, H. Takeda and K. Achiwa, J. Am. Chem. Soc., 1990, 112, 5876; J. M. Klunder, T. Onami and K. B. Sharpless, J. Org. Chem., 1989, 54, 1295.
- 6. U. Ader and M. P. Schneider, Tetrahedron: Asymmetry, 1992, 3, 521; H. S. Bevinakatti and A. A. Banerji, J. Org. Chem., 1991, 56, 5372; A. Kamal and M. V. Rao, Tetrahedron: Asymmetry, 1991, 2, 751; S.-H. Hsu, S.-S. Wu, Y. F. Wang and C. H. Wong, Tetrahedron Lett., 1990, 31, 6406.
- 7. A. Kamal, J. Org. Chem., 1991, 56, 2237.
- 8. A. Kamal, A. B. Rao and M.V. Rao, Tetrahedron Lett., 1992, 33, 4077.
- W. Boland, C. FroBl and M. Lorenz, Synthesis, 1991, 1049 and references therein; E. Dominguez, J. C. Carretero and A. F.- Mayorales and S. Conde, Tetrahedron Lett., 1991, 32, 5159; A.-H. Djerourou and L. Blanco, Tetrahedron Lett., 1991,32, 6325; A. B. Hajjar, P. F. Nicks and C. J. Knowles, Biotech. Lett., 1990, 12, 825; P. G. Hultin, F.-J. Mueseler and J. B. Jones, J. Org. Chem., 1991, 56, 5375; C. Bonini, P. Pucci, R. Racioppi and L. Viggiani, Tetrahedron Lett., 1992, 33, 29; C. Astorga, F.Robolledo and V. Gotor, Synthesis, 1991, 350; V. Gotor, R. Brieva, C. Gonzalez and F. Rebolledo, Tetrahedron Lett., 1991, 32, 9207.
- 10. The porcine pancreatic lipase (E.C. 3.1.1.3) employed was as purchased from Sigma, Type II crude.
- 11. A. Kamal, A. B. Rao and M. V. Rao, Biochem. Pharmacol., 1991,41, 1393.

IICT Communication No. 3087.